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MICROCELL DSC ANALYSIS OF SI-GRAPHITE ANODES PAIRED WITH HIGH ENERGY DENSITY CATHODES

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PROJECT OVERVIEW



Project Goal

Use **microcell safety testing** to understand how Si content impacts the **safety of Si/graphite anodes** when paired with an energy dense cathode and balance safety and performance.

Current Practice

Large format batteries subjected to destructive tests require larger quantities of materials and infrastructure to safely conduct, **slowing the rate at which testing proceeds.**

Why SNL?:

SNL has **established expertise in many forms of battery testing** and staff with experience in microcell safety testing.

Innovation

This work **bridges the gap** between small-scale electrode-only safety testing and full-cell testing, **improving the speed and efficiency of safety testing.** Earlier work from SNL has demonstrated the value of this approach.

Impact

These results will be interesting to: battery scientists, safety researchers, and battery manufacturers. This provides **foundational knowledge to improve the safety of two emerging battery chemistries.**

Alignment

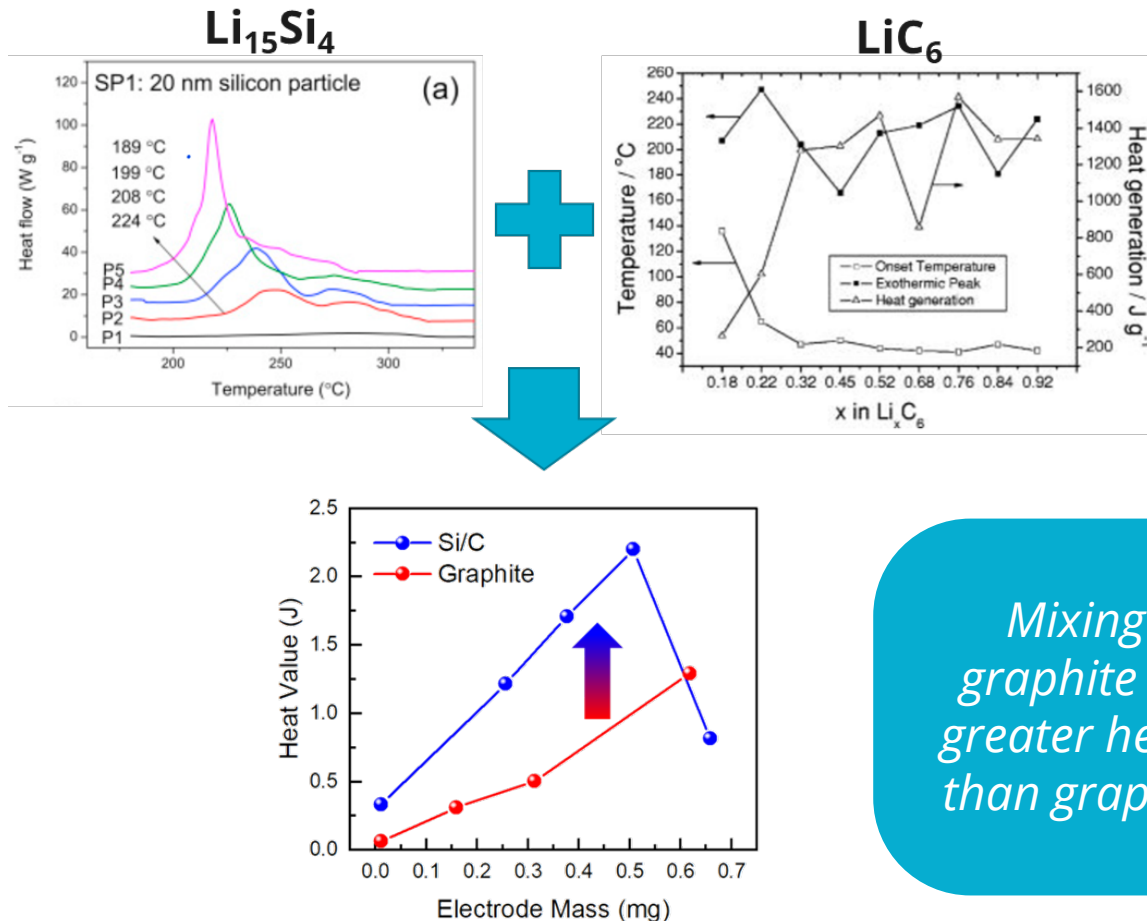
This **innovative** microcell approach improves our ability to study the safety of emerging battery chemistries, improving the **security** and **reliability** of future energy technologies. Inherently safer batteries reduce cost, allowing for the development of more **affordable** energy storage technologies.

SAFETY CONCERNS FOR SI/G + NMC811

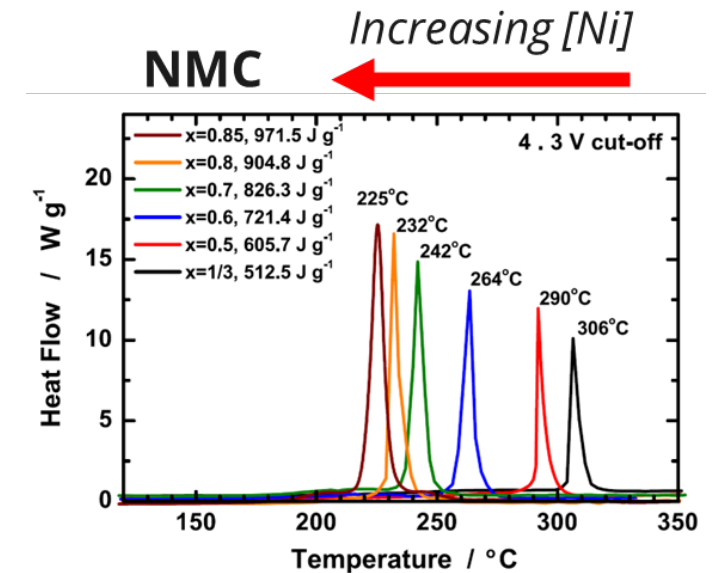


- Advanced materials like **silicon anodes** and Ni-rich $\text{LiNi}_{0.8}\text{Mn}_{0.1}\text{Co}_{0.1}\text{O}_2$ (**NMC811**) **cathodes** are promising high-energy density chemistries for next-generation batteries

The relative safety of these materials together remains unknown!



*Mixing Si with
graphite results in
greater heat release
than graphite alone*



*Increasing Ni content
increases heat flow
and decreases onset
temperature*



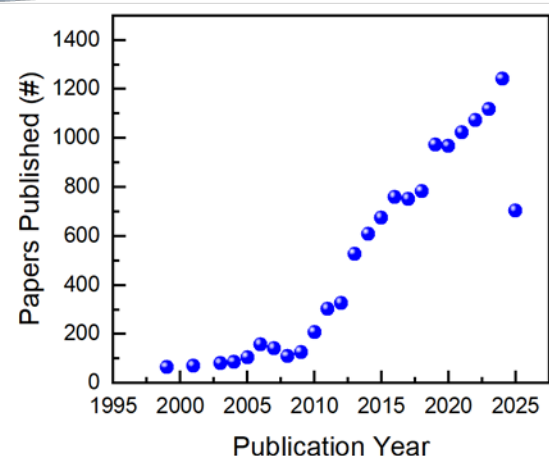
SI ANODES- A PROMISING GRAPHITE REPLACEMENT

Silicon anodes are an increasingly popular target for next-gen Li-based batteries for many applications

The Age of Silicon Is Here...for Batteries > The mainstay material of electronics is now yielding better energy storage

BY PRACHI PATEL | 04 MAY 2023 | 6 MIN READ

<https://spectrum.ieee.org/silicon-anode-battery>



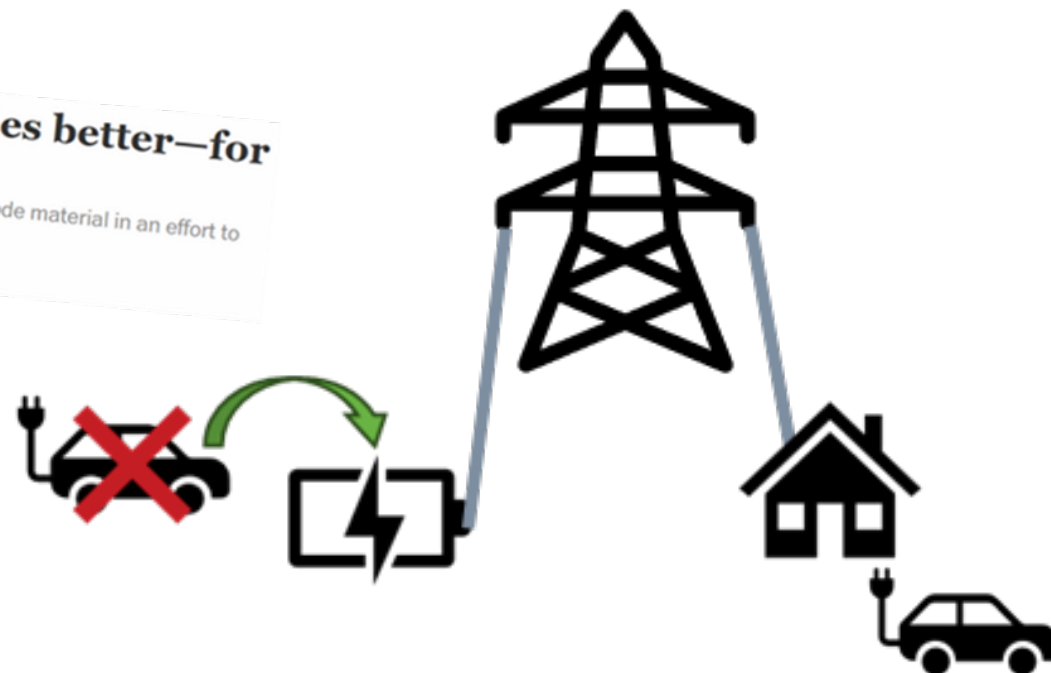
Silicon could make car batteries better—for a price

Several companies aim to lower the cost of the high-capacity anode material in an effort to dethrone graphite

by Matt Biola

February 4, 2024 | A version of this story appeared in Volume 102, Issue 4

<https://cen.acs.org/energy/energy-storage/Silicon-make-car-batteries-better/10204>



Batteries from multiple sources, and thus likely some with Si anodes, are likely to find their way to grid applications

NEWS ENERGY

EVs Are Essential Grid-Scale Storage > Before long, there will be more EV battery capacity than the grid can use

BY CHARLES Q. CHOI | 20 JAN 2023 | 2 MIN READ

Charles Q. Choi is a contributing editor for IEEE Spectrum.

<https://spectrum.ieee.org/electric-vehicle-grid-storage>

Article | [Open access](#) | Published: 16 May 2024

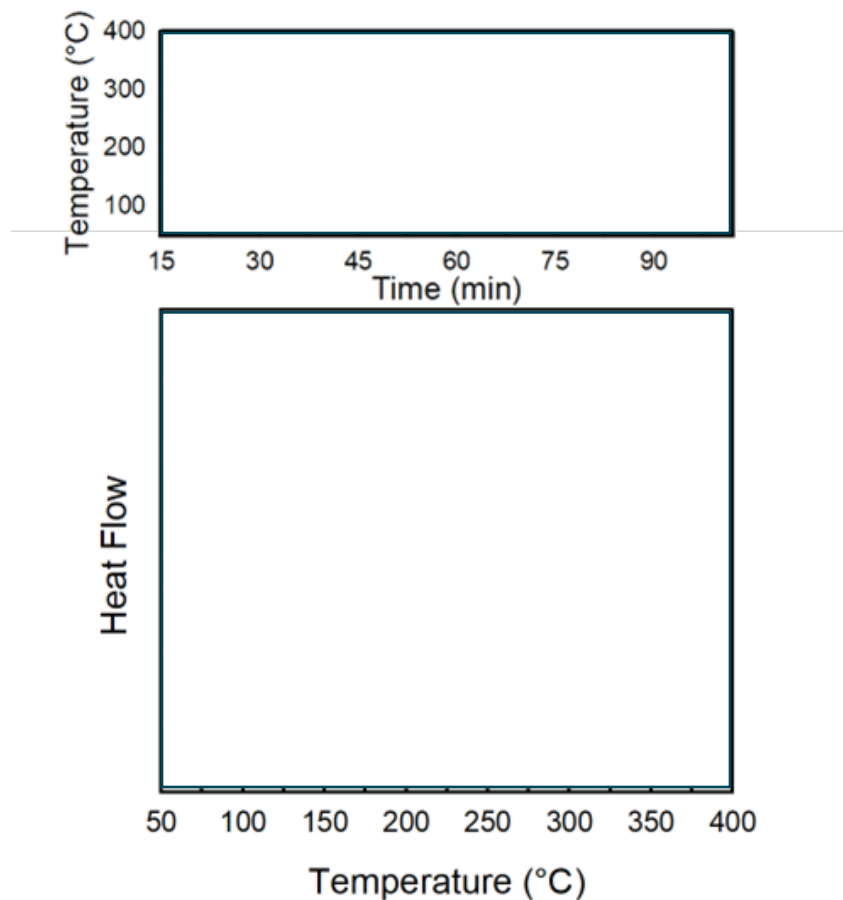
On the potential of vehicle-to-grid and second-life batteries to provide energy and material security

Fernando Aguilar Lopez, Dirk Lauinger, François Vuille & Daniel B. Müller

[Nature Communications](#) 15, Article number: 4179 (2024) | [Cite this article](#)

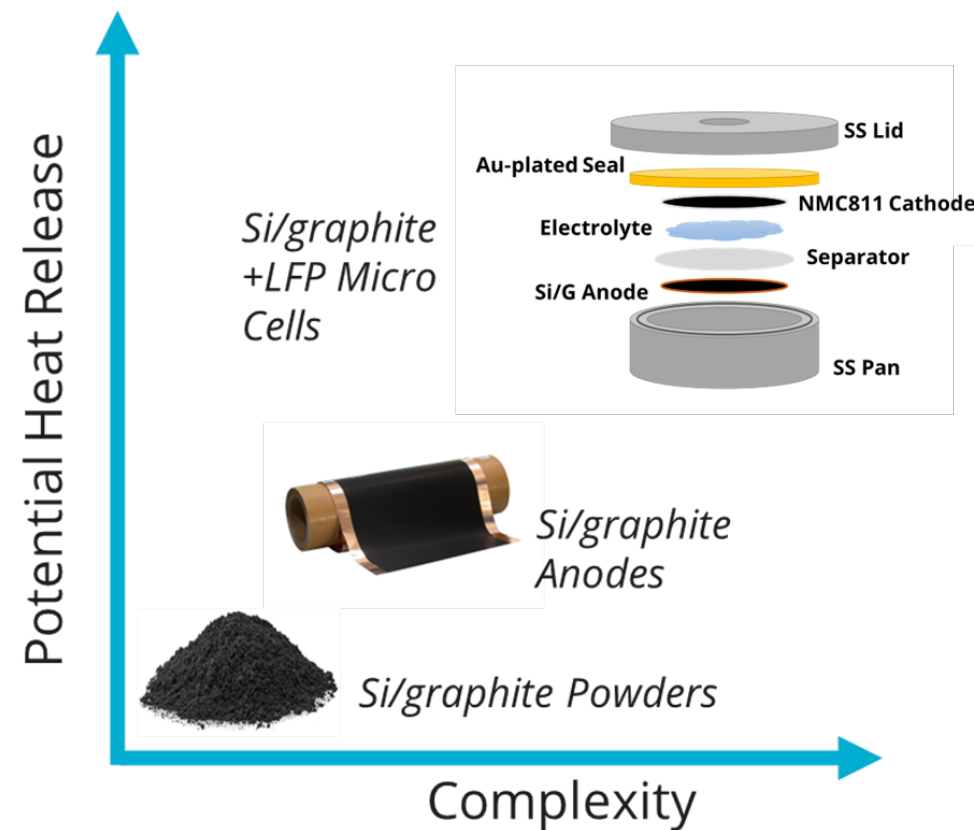
Aguilar F. et al. Nat Commun. 2024, 15, 4179

MATERIALS CHARACTERIZATION OF THE SAFETY OF SIG ANODES



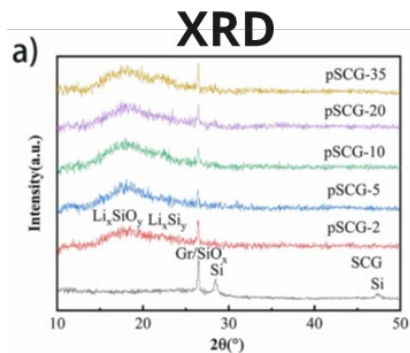
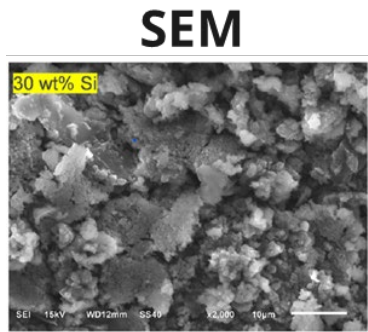
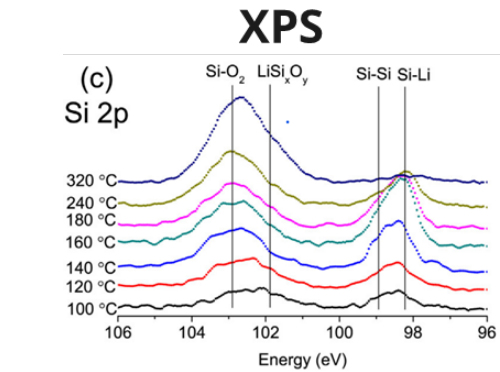
Differential Scanning Calorimetry (DSC)

Gradually ramp the sample temperature while measuring heat flow from the sample

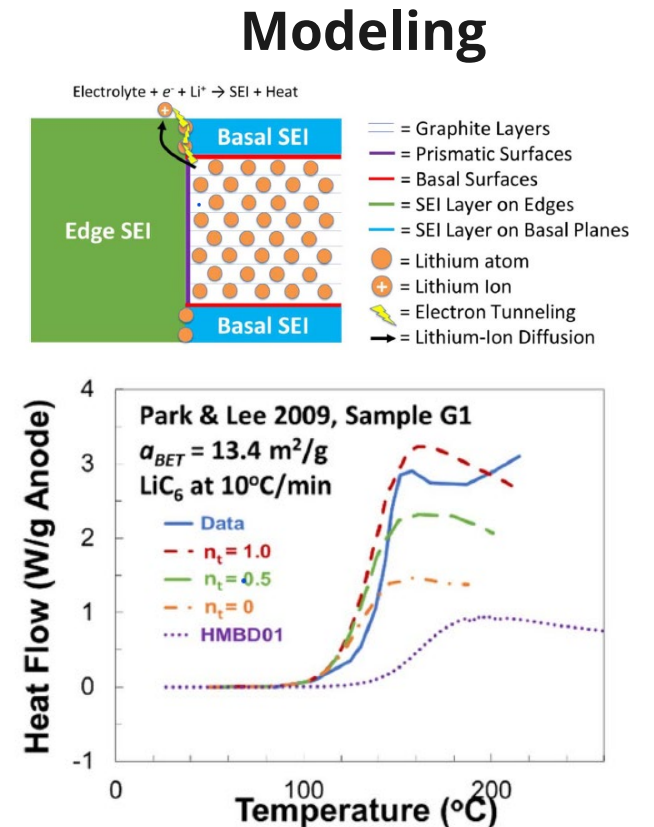
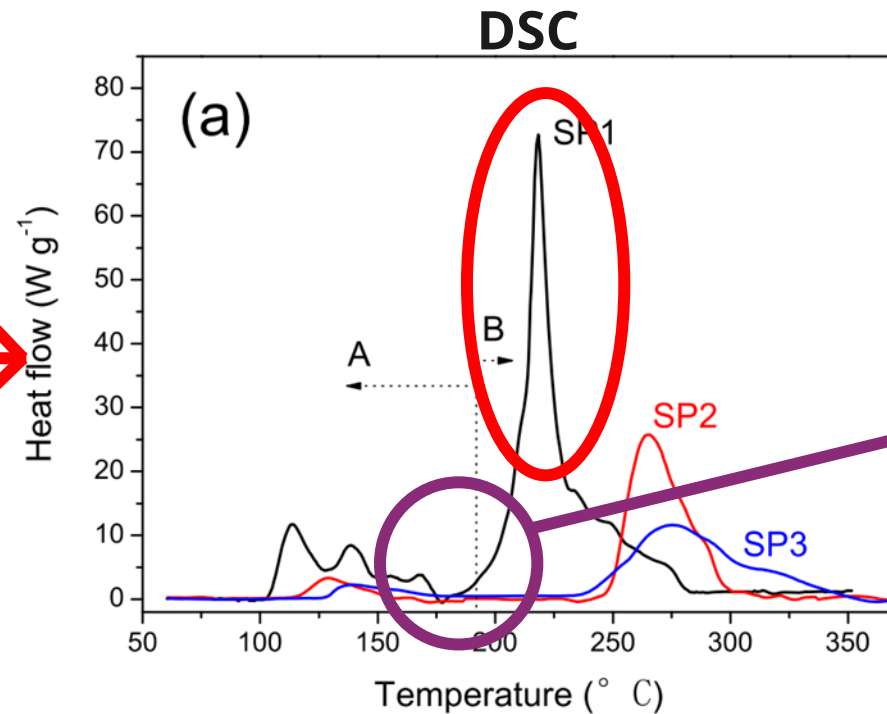


Increase sample complexity in order to better understand fundamental reactions

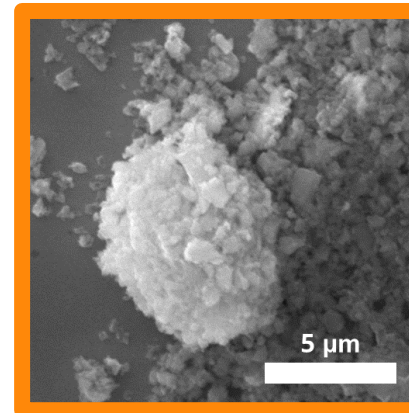
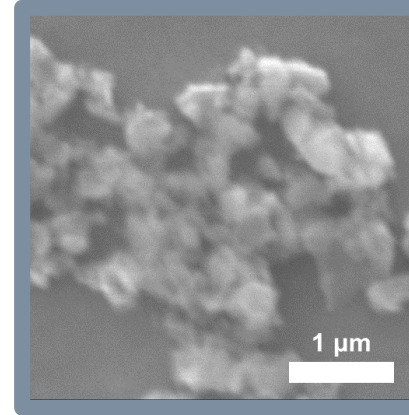
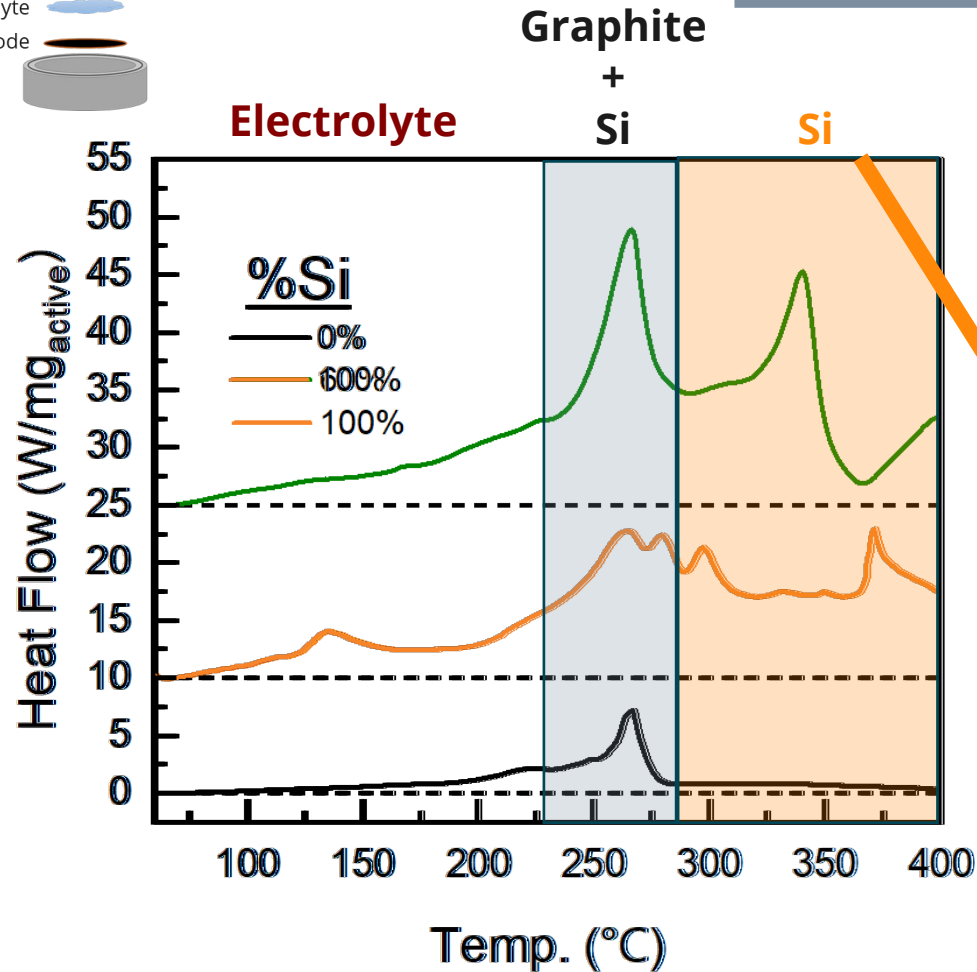
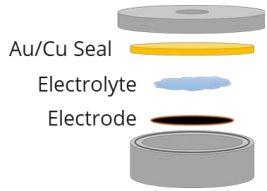
BOTTOM-UP MATERIALS CHARACTERIZATION OF THE SAFETY OF SILICON/GRAPHITE (SIG) ANODES



- Use pre- and post-DSC characterization as well as modeling to analyze *how* and *why* Si/G and NMC811 materials change

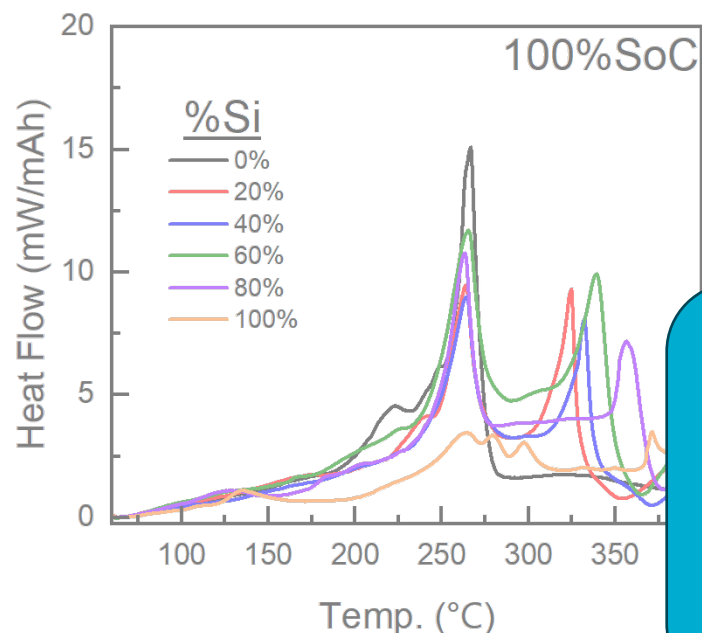


THERMAL RELEASE FROM ANODE-ONLY DSC



- 100%SoC anodes show similar thermal features largely related to the decomposition of **Li-Si** and **Li-C**
 - 230-290°C → **Li-C** + **Li-Si**
 - >290°C → **Li-Si**
- Prior studies suggest that at 100%SoC, larger Si particles yield higher degradation temperatures
- Mixed Si/graphite** DSC shows intermediate peaks between the temperature for pure Li-C or Li-Si decomposition

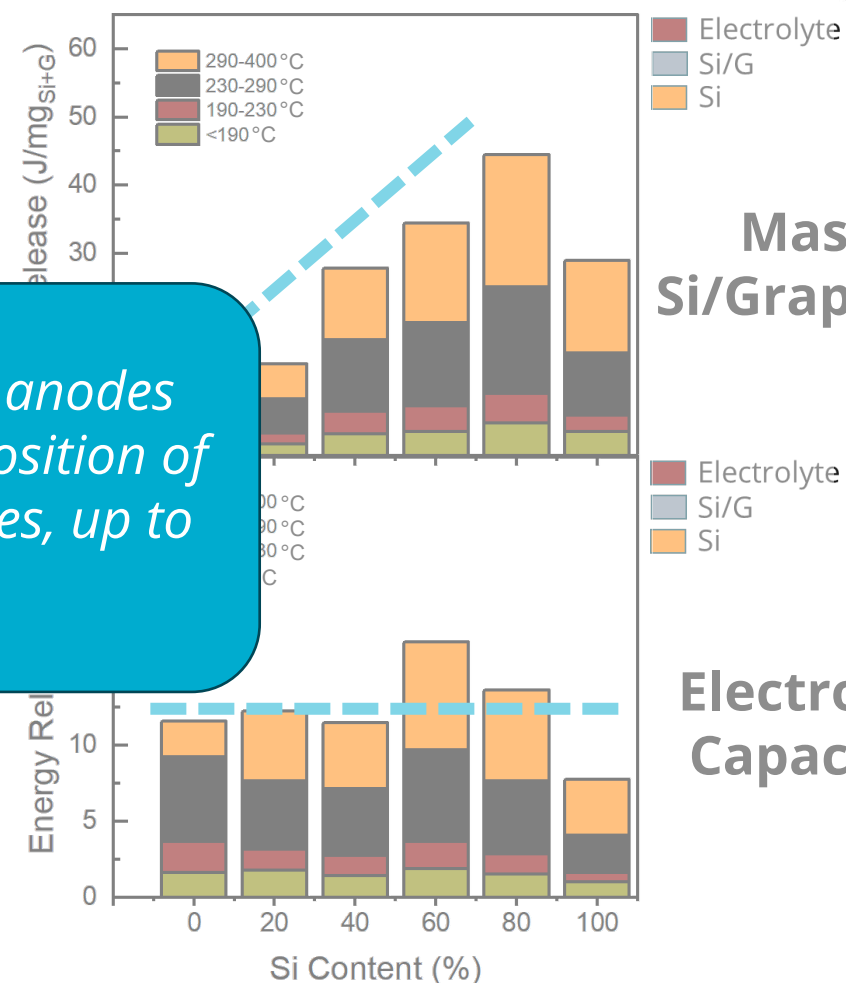
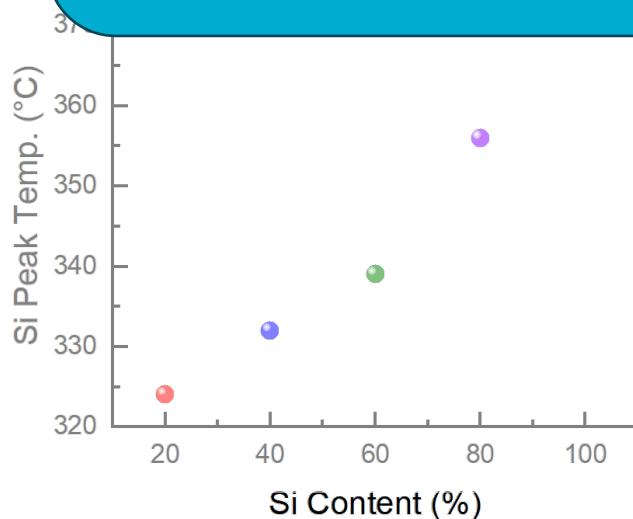
THERMAL RELEASE FROM ANODE-ONLY DSC



- Mixed Si/graphite anodes **all** exhibit a sharp exotherm from Li-Si and Li-C decomposition at ~265 °C

In absence of a cathode, charged anodes release more energy from decomposition of Li-Si and Li-C as Si content increases, up to 80 wt.% Si

- Increasing Si content drives decomposition peak to higher temperatures
- Si agglomerates may be diluted by graphite, resulting in less “bulk” behavior at low Si content

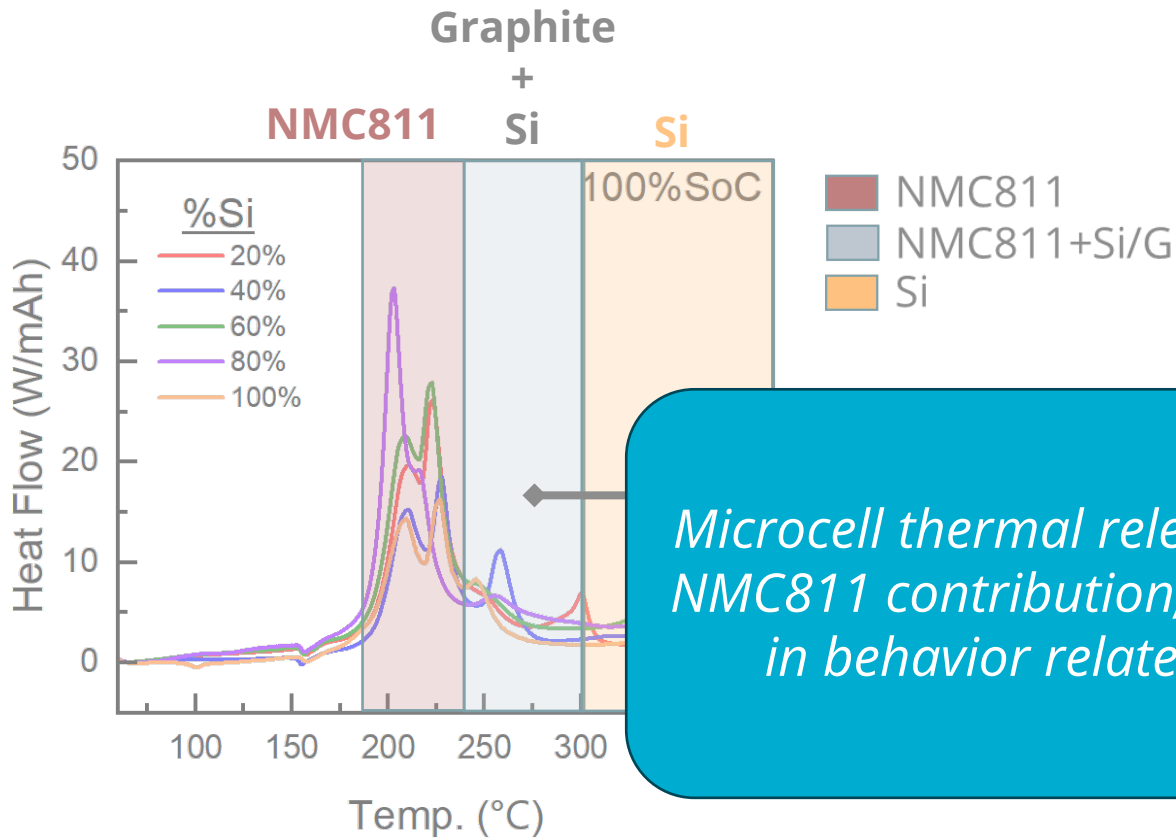


Mass
Si/Graphite

Electrode
Capacity

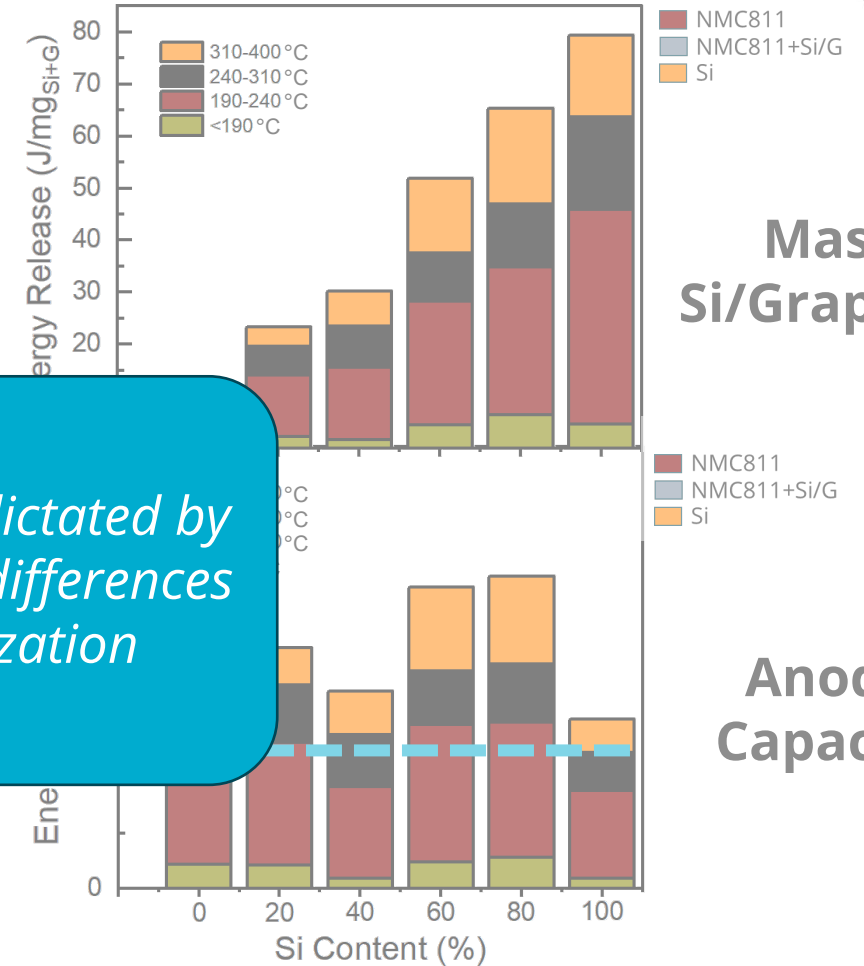
- How you normalize and interpret your energy release matters.
- Trends differ between J/mg and J/mAh

THERMAL RELEASE FROM MICROCELLS



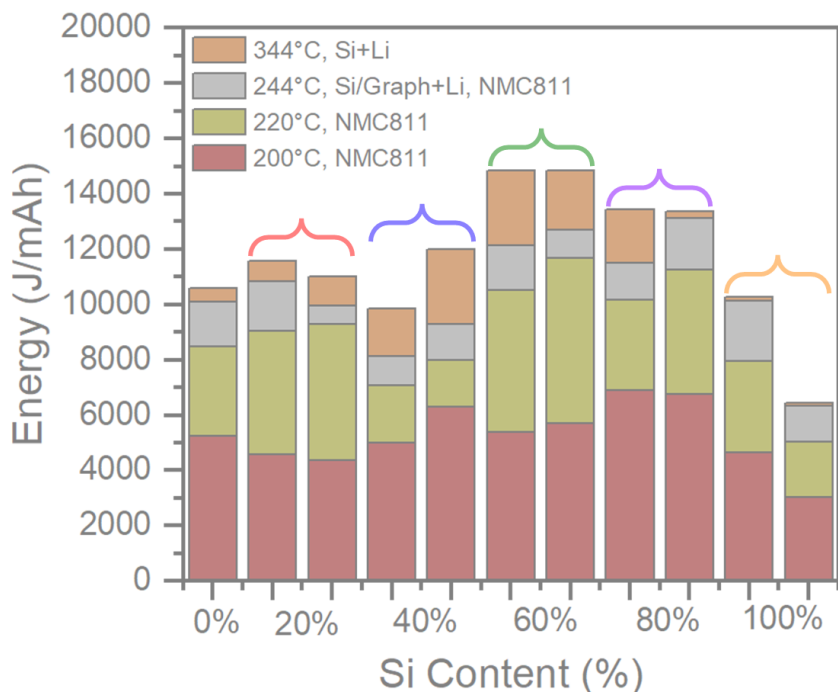
Microcell thermal release largely dictated by NMC811 contribution, with some differences in behavior related to normalization

- 100%SoC microcell behavior dominated by **NMC811** decomposition below 240 °C
- Smaller peaks >240 °C appear to correspond to the **Si**-induced doublets from the anode, but are more uniform



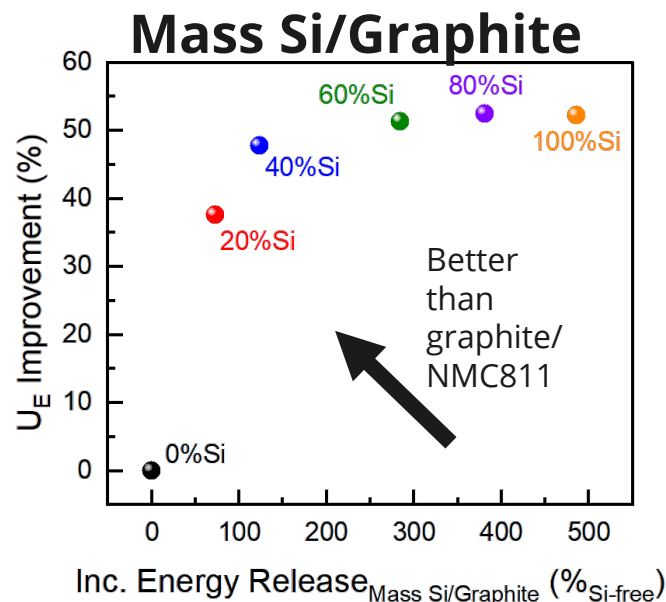
- Total energy released still trends upwards with the quantity of Si
- Energy release from microcells appears to be more dependent on capacity than anode-only

REPRODUCIBILITY OF RESULTS AND SAFETY IMPLICATIONS



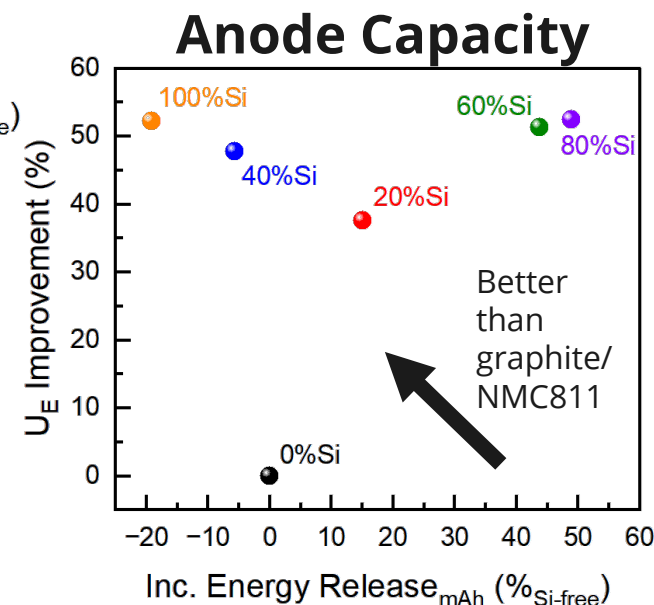
- Replicate microcells with Si/graphite mixtures tend to have relatively good reproducibility
- Relative ratios tend to be conserved for various peaks

Microcells tend to have fairly reproducible DSC results



Normalization matters, but ~20-40%Si shows best compromise

- Total energy released can be used as yardstick for overall safety, allowing comparison based on potential improvement to energy density (U_E)

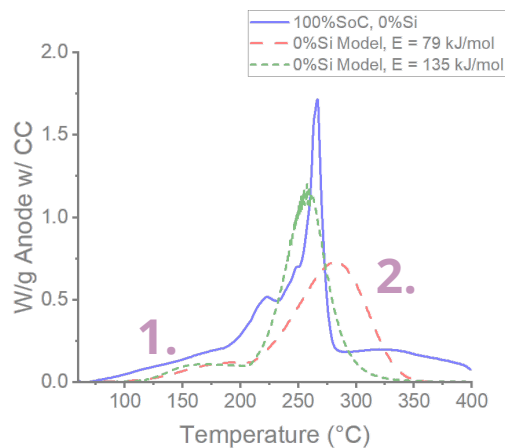


MODELING OF SI/GRAPHITE ANODES AND NMC811 CATHODES



Graphite Anode

(Confirmation of Existing Model)



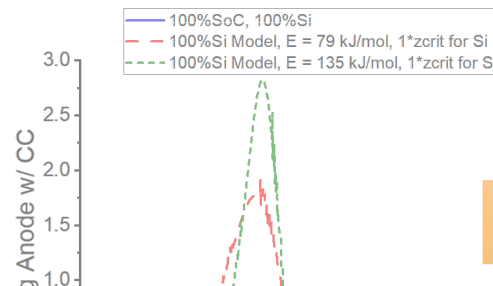
1.
Prediction of low
initial rate

2.
Predicts high rate onset
and total heat

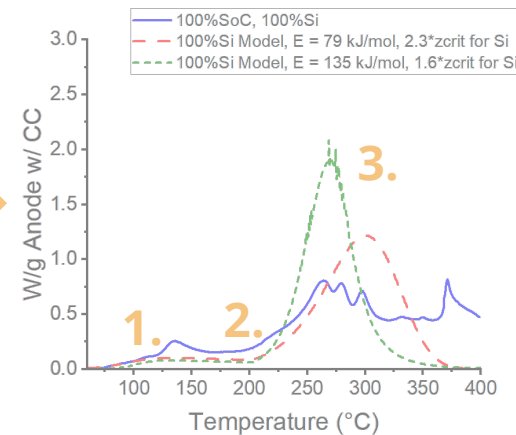
repre
unde
t
act

Si Anode

(1 parameter changed from graphite model)



Optimize

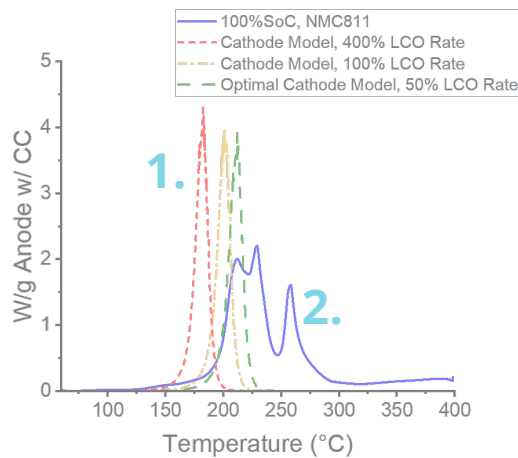


2.
tunneling
increased
atch higher
rate onset

3.
Lower activation
energy provides
better match for Si

NMC811 Cathode

(1 parameter changed from LCO model)



NM
electrolyte slower
than LCO

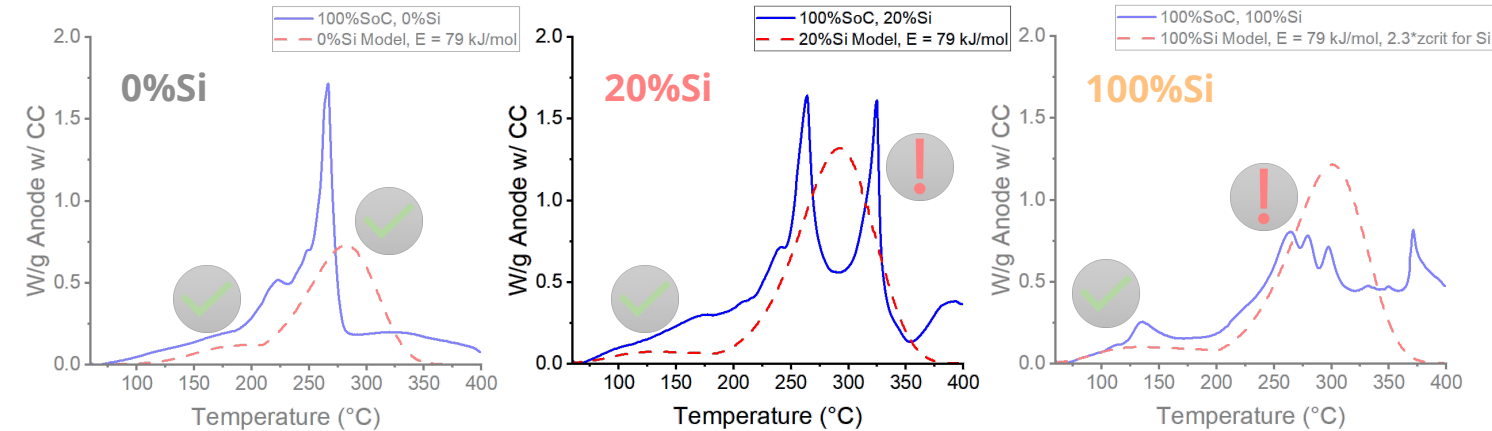
2.
Secondary/tertiary
cathode peaks not
modeled, but total heat
from all peaks
adequately predicted

Graphite, Si, and NMC811 DSC behavior can
be adequately captured by existing models

MODELING OF SI/GRAPHITE ANODES AND MICROCELLS



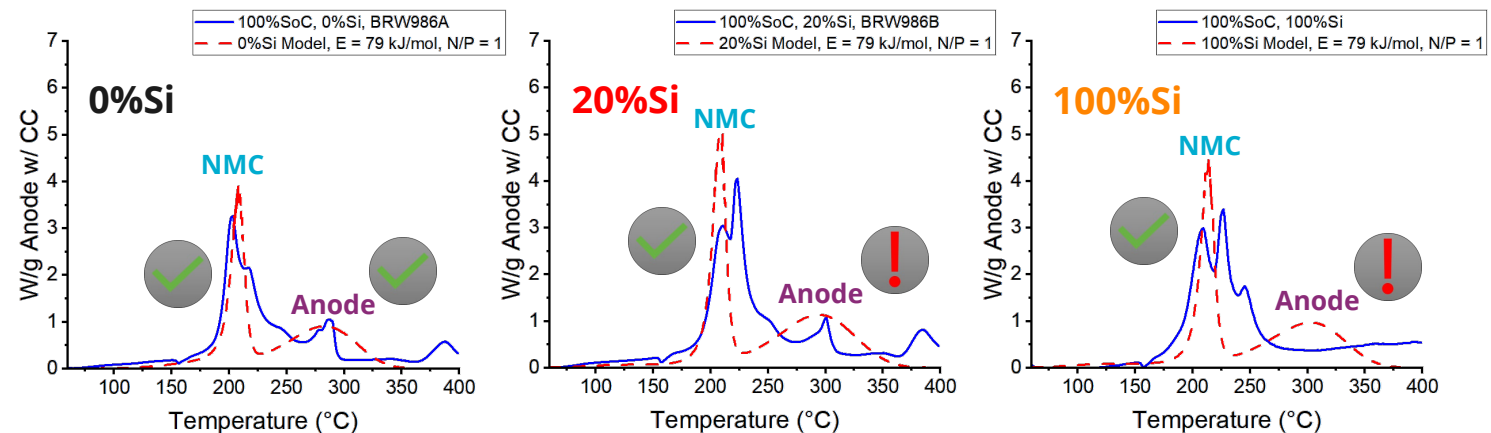
Anodes Only



- Prediction of slow initial rate consistent for both graphite and Si anodes
- Onset of higher rates well predicted, **but** multiple peaks appear to be more common with Si present
 - Model only designed for single peak.

- Behavior of cathode appears well predicted for all microcells
- Continue to see two anode peaks in experimental data; model with one peak overpredicts max anode rate
- Predicts total heat release of anode peak

Microcells



Mixture of Si and Graphite results in unexpected deviations from modeled behavior in anode that can be explored by changing the model

CONCLUSIONS AND FUTURE DIRECTIONS (FY26+)

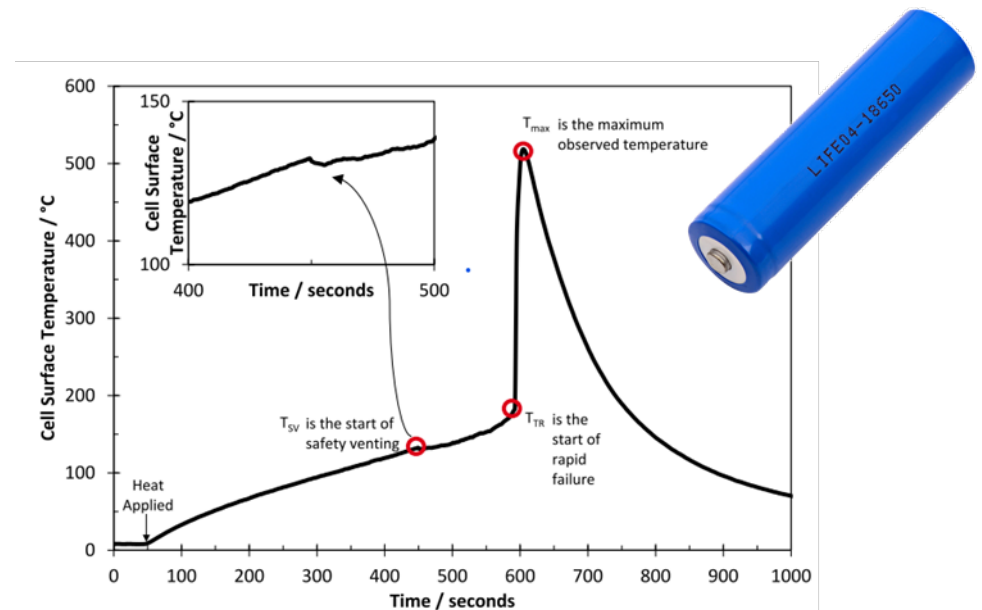


Conclusions

1. Mixture of Si and graphite *does* impact heat release, likely through Si agglomerate size
2. Most microcell energy released during NMC811 degradation, but *is* influenced by Si content
3. 20-40% Si best balances battery safety and performance for Si/graphite anodes paired with NMC811 cathodes

Future Directions

- Larger scale (18650) safety tests using selected Si/graphite mixtures



ACKNOWLEDGEMENTS



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- Dr. Randy Shurtz

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- Dr. Nathan Johnson (DSC)

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U.S. DEPARTMENT
of **ENERGY**

Office of
Electricity

Energy Storage



THANK YOU