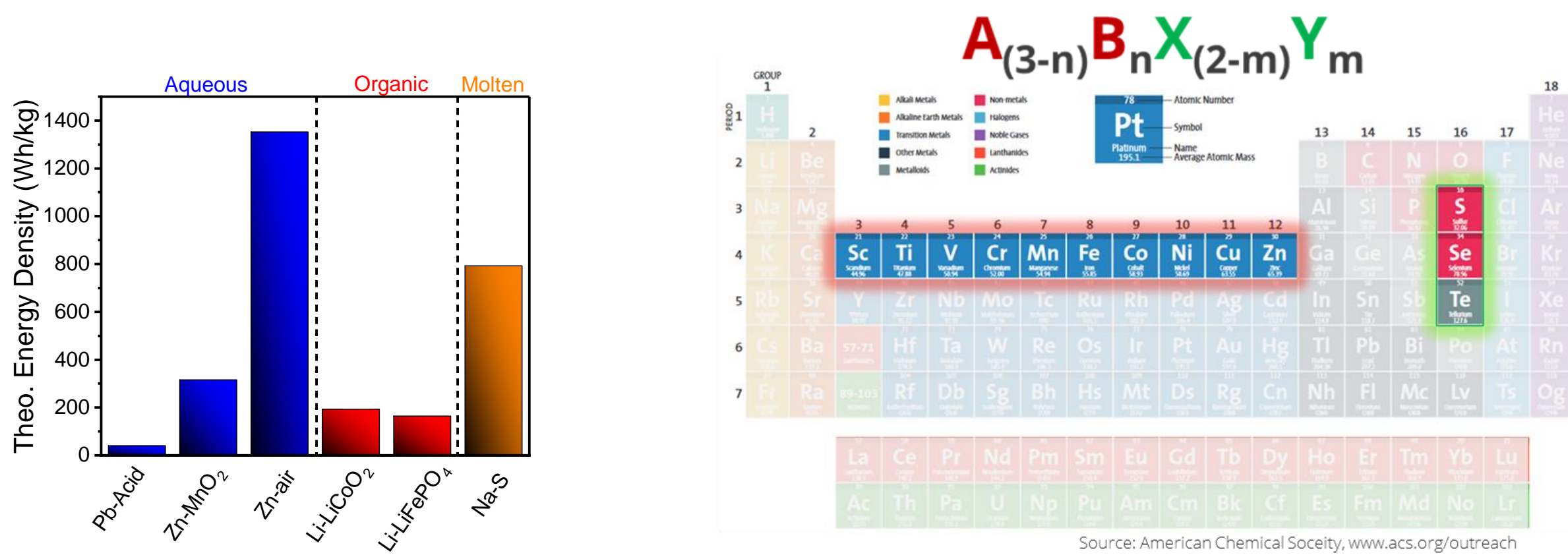


TRANSITION METAL MULTICHALCOGENIDES AS BIFUNCTIONAL OXYGEN ELECTROCATALYSTS FOR ZINC-AIR BATTERIES

Bryan R. Wygant,^{1*} Ciara N. Wright,² Brian A. Washington,⁴ Gabriel A. Goenaga-Jimenez,⁴ Thomas A. Zawodzinski, Jr.,^{4,5} Timothy N. Lambert^{2,3†}

¹Nanoscale Sciences, ²Department of Photovoltaics & Materials Technologies, ³Center for Integrated Nanotechnologies, Sandia National Laboratories, Albuquerque, New Mexico 87185, USA, ⁴Department of Chemical & Biomolecular Engineering, The University of Tennessee Knoxville, Knoxville, Tennessee 37996, United States, ⁵Oak Ridge National Laboratory, Oak Ridge, Tennessee 37830, United States
*bwygant@sandia.gov, †tnlambe@sandia.gov

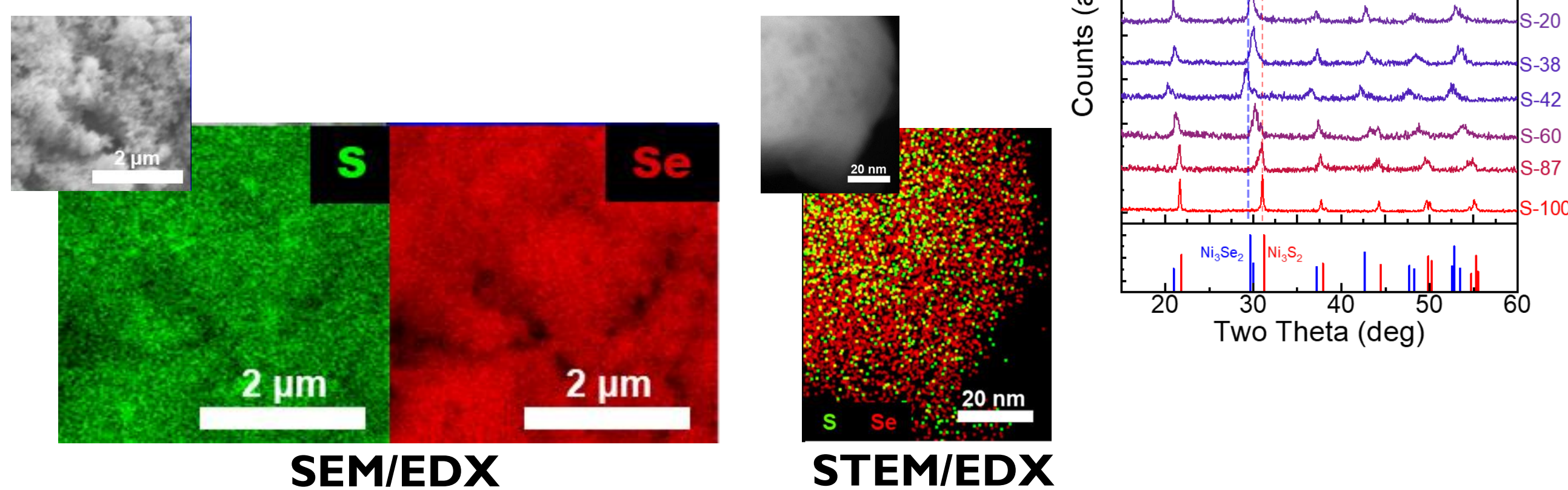
Background



- Zn-air (and other metal-air chemistries) are viable alternatives to Li-ion batteries, but require better bifunctional oxygen electrocatalysts (BOEs) to replace precious metals like Pt and Ru
- Transition metal chalcogenides have demonstrated good performance as oxygen evolution reaction (OER) and oxygen reduction reaction (ORR) electrocatalysts and have a broad degree of chemical "flexibility", making them excellent targets for further research

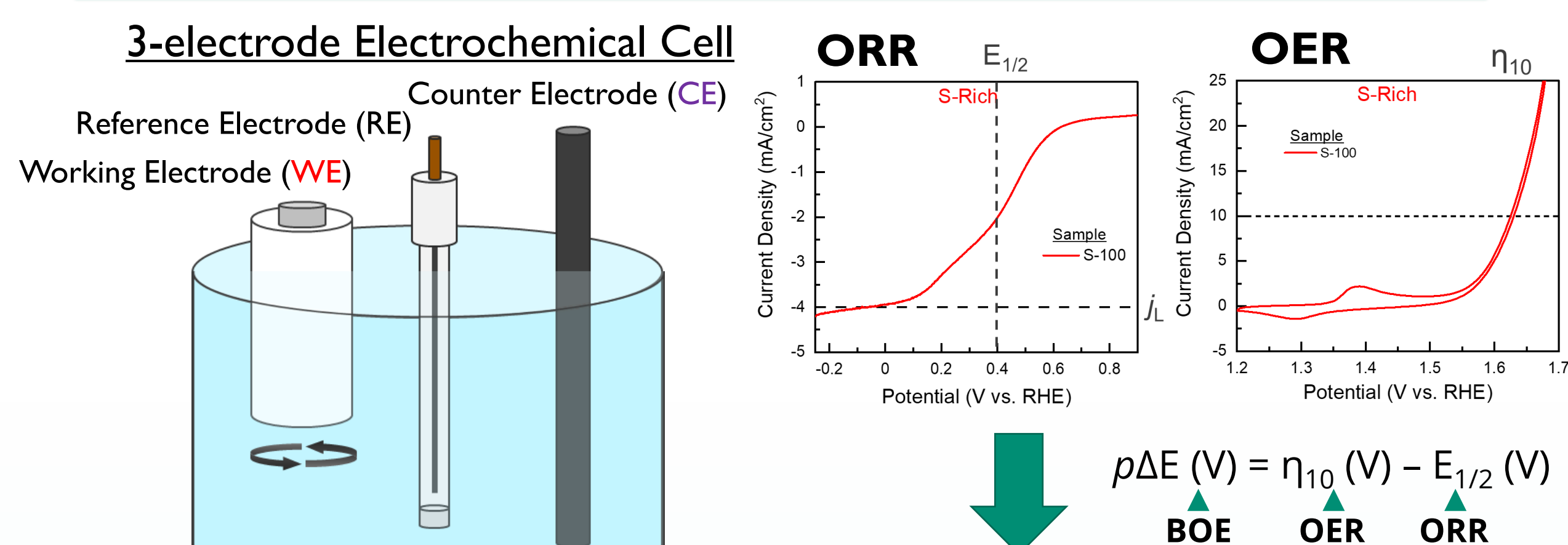
Synthesis and Characterization

- Hydrothermal reaction to produce series of $Ni_3S_{2-x}Se_x$ (NiSSe) powders

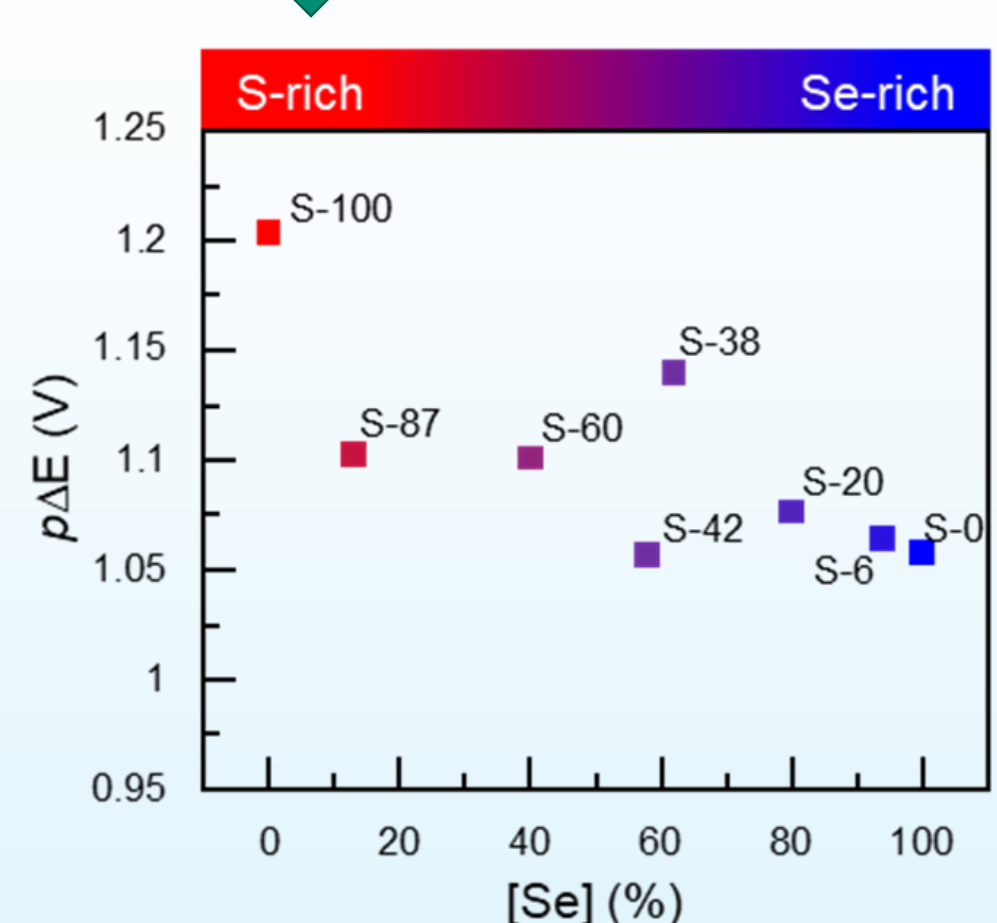


- Characterization shows that the materials show a transition from Ni_3S_2 to Ni_3Se_2 , but that the S/Se in the material is evenly distributed within discrete particles

Electrocatalytic Testing

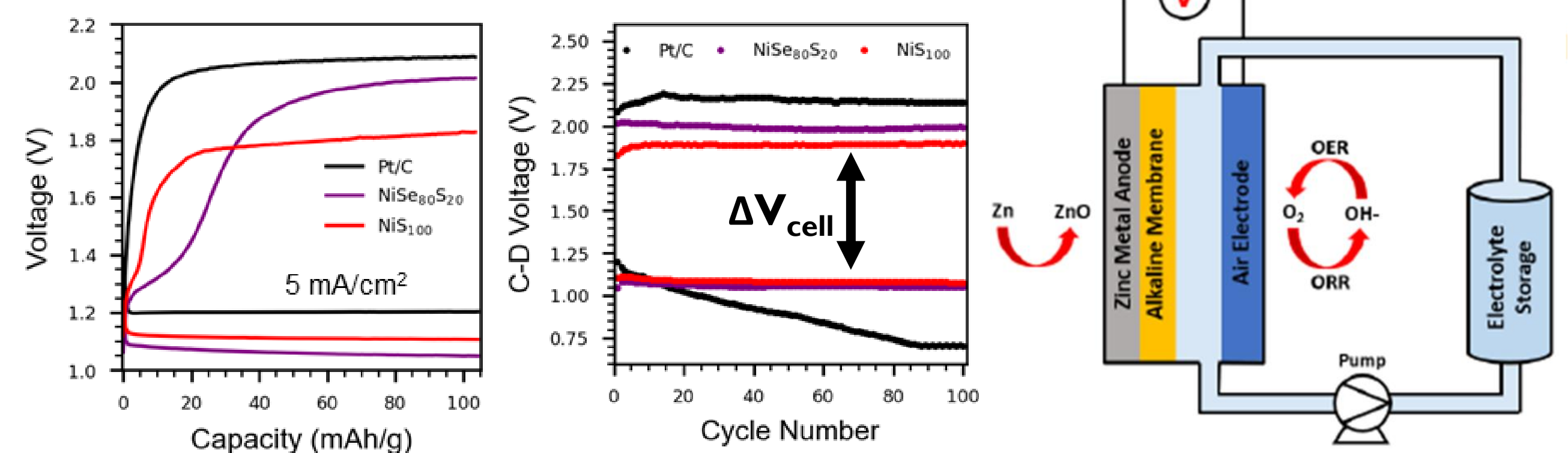


- Electrocatalytic testing in 3-electrode cells allows testing for OER and ORR performance, as well as determination of $p\Delta E$, a measure of BOE performance

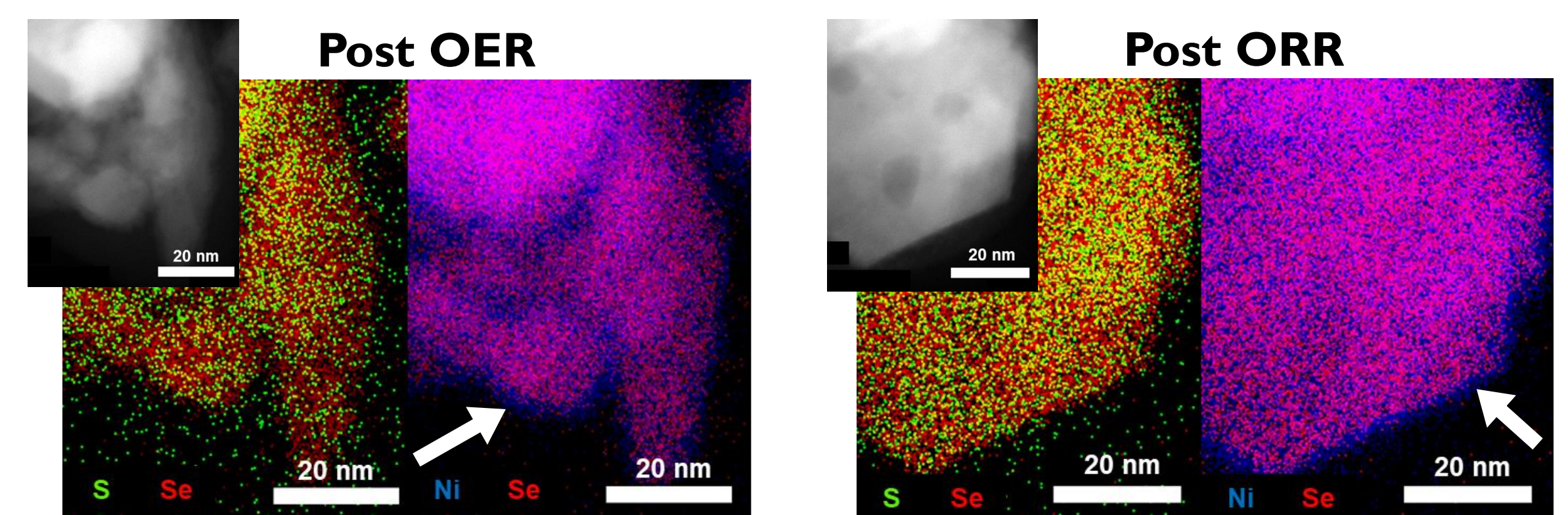


Zn-Air Battery Performance

- Testing in a flow Zn-air battery shows improved BOE performance for NiSSe compared to commercial Pt/C



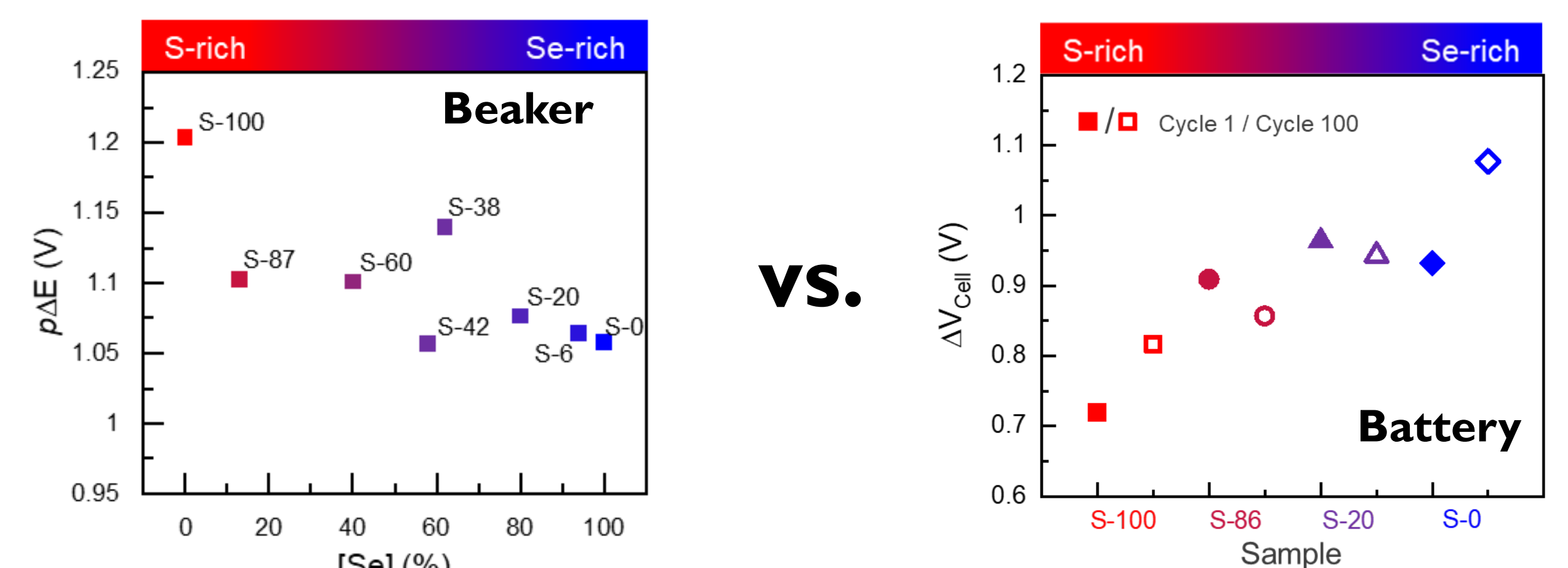
Controlled Surface Oxidation



- STEM EDX mapping shows that the surface of NiSSe after OER and ORR consists of a thin O-rich layer atop the S/Se bulk, likely active electrocatalyst
- S is more readily oxidized, Se is more stable and more conductive; mixing them helps generate optimal material for BOE

Differences in Beaker vs. Battery

- For limited tests in batteries, the trend for overall BOE performance is inverted; Se-rich appear to be worse BOEs



Conclusions + Future Directions

- Hydrothermal synthesis is a simple, tunable method for producing mixed metal chalcogenides
- Electrocatalytic performance of NiSSe materials is promising, and suggests that Se-rich materials may be suitable BOEs for Zn-air batteries

BUT

- A better understanding of why BOE trends differ for 3-electrode tests and batteries is needed to help drive further research
- Exploration of more earth-abundant metals (e.g., Fe) will help further improve costs for developing and building Zn-air batteries

Wygant et al., ACS Catal., 2023, 13, 9245–9253